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WORKSHEET Ionic Bonds Use with

Chapter 7, Section 7.2 1. How many

valence electrons does a neutral

magnesium (Mg) atom have? 2. What is

the charge on a magnesium ion? What

does magnesium have to do to form such

an ion, and why does it tend to do so? 3.

How many valence electrons does a single

neutral chlorine (Cl) atom have? 4.

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Chemistry: Matter and Change Chapter 5 .

uest 11 Name: Date: Hour: Information:

Energy Levels and Sublevels ...

Considering your answer to question 5, how many electrons can fit in the entire 5t energy level? can (AVel is a ha AC5a)

50đ 7. Write the notation for an electron spinning clockwise in a p sublevel ...

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WORKSHEET The Activity Series Use with Chapter 9, Section 9.2 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

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Chemistry Answers Chapter 19 Author:  
s2.kora.com-2020-11-30T00:00:00+00:01  
Subject: Teaching Transparency  
Worksheets Chemistry Answers Chapter  
19 Keywords: teaching, transparency,  
worksheets, chemistry, answers, chapter,  
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Chemistry Answers Chapter 19  
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8 . 1.7, the CF bond is polar covalent. (b)  
!ENCI"P = ENCI " ENP. = 3.0" 2.1  
!ENCI"P = 0.9. Since  $0.5 < EN. A. < 1.7$ ,  
the PCl bond is polar covalent. (c)  
!ENCI"Li Section 8.5 Electronegativity  
and Polarity 265 Compare and contrast  
polar . that is in excess.

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WORKSHEET Atomic Orbitals 1. What is

the shape of an s orbital? Class 16 Use

with Chapter 5, Section 5.2 2. What is the

relationship between the size of an s

orbital and the principal energy level in

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which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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