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hybridization of the central carbon in  $\text{CH}_3\text{C}\equiv\text{N}$  and the bond angle  $\text{CCN}$  are a.  $\text{sp}^2$ ,  $180^\circ$ . b.  $\text{sp}$ ,  $180^\circ$ . c.  $\text{sp}^2$ ,  $120^\circ$ . d.  $\text{sp}^3$ ,  $109^\circ$ . 2. Which of the following statements about an  $\text{sp}$  hybridized carbon is FALSE?

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