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CHAPTER 7 REVIEW Chemical
Formulas and Chemical

Compounds SECTION 1 SHORT

ANSWER Answer the following
questions in the space

provided. 1. c In a Stock
system name such as

iron(III) sulfate, the Roman
numeral tells us (a) how

many atoms of Fe are in one
formula unit. (b) how many

sulfate ions can be attached
to the iron atom. (c) the

charge on each Fe ion.

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unit of a molecule or ion is
the sum of the average
atomic masses of all atoms
represented in its formula.
Percentage composition.

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Compounds SECTION 3 SHORT
ANSWER Answer the following
questions in the space
provided. 1. Label each of
the following statements as
True or False: True a.

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ions. ions from page 209 and
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molecule, formula unit, or
ion is the sum of the
average atomic masses of all
atoms represented in its
formula. a compound's molar
mass is numerically ...
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Gravity. Created by.
dolphinking. study the
thing. Terms in this set
(25) Chemical formula.
indicates the relative
number of atoms of each kind
in a chemical compound.
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