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equation.

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Solve each of the following problems.

Assume excess of any reactant that isn't mentioned, unless otherwise specified.

Show your work, including proper units, to earn full credit. 1.

Given the balanced equation, show what the following molar ratios (i.e., the conversion

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factors) should be.

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Data Analysis Data

Analysis 1. A sample of

aluminum is placed in a

25-mL graduated

cylinder containing 10.0

mL of water. The level

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of water rises to 18.0  
mL. Aluminum has a  
density of 2.7 g/mL.  
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Calculate the mass of  
the sample. 2. Saturn is  
about 1 429 000 km  
from the Sun.

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$4.93 \times 10^{-5} \text{ L}$  or  $49.3$

$\mu\text{L}$  In Example 12.2.1  
and Example 12.2.2, the  
identity of the limiting  
reactant has been

apparent:  $[\text{Au}(\text{CN})_2]^-$

,  $\text{LaCl}_3$ , ethanol, and  
para -nitrophenol.

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