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Acids Bases and Salts QuizACIDS BASES \u0026 SALTS-FULL CHAPTER || CLASS 10 CBSE CHEMISTRY Acid Bases \u0026 salts chapter 5 QUESTION ANSWERS full explaination class 7th ncert Science NCERT Solutions (Part 2) - Acid, Bases And Salts | Class 10 Chemistry

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Solution: The reaction of the acid + base gives a product of salt + water, which is considered as neutralization reaction. Examples: NaOH + HCl NaCl + H 2 O

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Ans:- Salts are classified into four categories on the basis of the strength of acid and base. they are: Salt of strong acid and strong base.e.g NaCl, KCl; Salt of strong acid and weak base. e.g. NH 4 Cl, NH 4 I; Salt of weak acid and strong base. e.g. CH 3 COONa, COO Na + Salt of weak acid and weak base. e.g. CH 3 COO NH 4 + How is common salt found in nature?

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2. Bases react with . acids to produce salts and water; water to produce acids and salts; salts to produce acids and water; neither acids, salts, nor water

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The salts of strong acids and weak bases give acidic solution having pH less than 7. Example, NH 4 Cl, Ammonium Chloride will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example, Na 2 CO 3, Sodium Carbonate will have pH more than 7. Question 7.

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Answer A reaction in which an acid and base react with each other to give a salt and water is termed as neutralization reaction.For Example: (i) NaOH + HCl NaCl + H 2 O (ii) HNO 3 + KOH KNO 3 + H 2 O 15. Give two important uses of washing soda and baking soda. Answer Two important uses of washing soda are:

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Answer. (a) HCl is acid and NaOH is base whose combination forms the common salt. Its formula is NaCl (Sodium chloride). It is obtained from sea water. (b) Rock salt is the common name for the mineral halite. Its chemical formula is NaCl. It may be white or light blue or yellow depending upon impurities present in it.

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(A) inorganic acid (B) organic acid (C) soft organic acid (D) strong inorganic acid. Answer. Answer: (a) 5. When milk of magnesia reacts with acetic acid it produces (A) basic salts (B) acidic salts (C) neutral salt (D) complex salt. Answer. Answer: (c) 6. Which of the following phenomena will occur when a small amount of acid is added to water ...

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Turmeric is a natural indicator. On adding its paste to acid and base separately, which colours would be observed. (a) Yellow in both acid and base (b) Yellow in acid and red in base (c) Pink in acid and yellow in base (d) Red in acid and blue in base. Answer. Answer: (b) Yellow in acid and red in base

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Acids, bases and salts An acid is a compound that is defined by its physical and chemical properties. Acids taste sour and react with metals and polyatomic ions called carbonates. A carbonate is a charged cluster of Carbon and Oxygen atoms.

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Acid + Base ---> Salt + Water. Question 3: Rain becomes acidic due to. carbon dioxide. sulphur dioxide. nitrogen dioxide. all of the above. Correct Option is : 4. Solution : Rain becomes acidic because of carbon dioxide, sulphur dioxde and nitrogen dioxide. They dissolve in rain drops to form carbonic acid, sulphuric acid and nitric acid ...

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