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Notes
Question 1.

Calculate the radius and velocity of the electron in the n th orbit in hydrogen atom from the following data,
 $e = 1.6 \times 10^{-19}$.
 $C, m = 9.11 \times 10^{-31}$, $h = 6.625 \times 10^{-34}$ JS, ? 0

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 $= 8.85 \times 10^{-12}$
 Fm^{-1} and $c = 3 \times$
 $10^8 \text{ s}^{-1} \dots$

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and molecules,
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interaction of
each atom or ...
Notes
than the size of
the nucleus (see
Chapter 11,
Section 11.6 in
Class XI Physics
textbook). Thus,
the electrons
would seem to be
at a distance
... second law
of motion and
the Coulomb's

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Questions and
Answers.

Question 1. (a)
Two stable
isotopes of

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Lithium have
respective
Notes

abundances of
7.5% and 92.5%.

These isotopes
have masses

6.01512 u and

7.01600 u,

respectively.

Find the atomic
mass of lithium.

(b) Boron has
two stable
isotopes, .

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short answer
questions in
Part A, short
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and long answer
questions or ...

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challenging the
brain to think
greater than
before and
faster can be
undergone by
some ways.

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experience, Chapter

adventuring,

studying,

training, and

more practical

goings-on may

back you to

improve. But

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realize not have

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to acquire the

business ...

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Magnetism and

Matter Three

Marks Questions

and Answers.

Question 1.

Mention the

range of

magnetic

susceptibility

for (a)

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Diamagnetic Chapter

substances (DMS)

(b) Paramagnetic
substances (PMS)

(c)

Ferromagnetic
substances

(FMS). Answer:

(a) For DMS, χ
is very low and
negative - 10^{-9}
to -10^{-5} . e.g :
For copper, χ is
 -9.8×10^{-6} .

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